



## Original Paper

# CT investigation on oilwell cement deterioration caused by H<sub>2</sub>S along a leaking channel under high temperature: Insights for geothermal applications



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## ABSTRACT

The alteration of oilwell cement due to H<sub>2</sub>S poses a significant threat to wellbore structural integrity in geothermal environments. However, laboratory studies on the cement deterioration process caused by H<sub>2</sub>S flow along a leaking channel under high-temperature conditions remain scarce. In this study, computed tomography (CT) scanning was utilized to assess the morphological changes and alteration patterns of oilwell cement caused by H<sub>2</sub>S flow in multiple dimensions. Additionally, scanning electron microscopy (SEM) coupled with energy-dispersive X-ray spectroscopy (EDS) and Fourier transform infrared spectroscopy (FTIR) were applied to elucidate the microscale mechanisms responsible for the H<sub>2</sub>S-driven alteration. The results show that: H<sub>2</sub>S flow along the cement channel results in increased cement matrix porosity and formation of large pores, which are especially evident in regions adjacent to the channel. Chemical etching and secondary crystal growth contribute to the expansion of channel dimension and roughening of the channel wall. Consequently, the permeability of the cement matrix exhibited a marked increase of 45% over a period of 14 days. At the microstructural level, compared to unaltered oilwell cement, which exhibits a homogeneous texture and fine particle composition, exposure to H<sub>2</sub>S leads to the formation of a heterogeneous and fractured structure within the cement. As a result of sulfidation reactions, a surface layer approximately 1 mm in thickness forms on the cement, which is depleted in calcium and enriched in silicon. The identification of metallic sulfides elucidated the chemical mechanisms responsible for the deterioration of cement properties. In summary, the flow of H<sub>2</sub>S through the channel within the cement causes significant alteration of the cement structure compared to other alteration modes.

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## 1. Introduction

Oilwell cement is a critical sealing material used in wellbore construction for geothermal energy extraction, petroleum production, and geological carbon sequestration projects, where

maintaining long-term structural integrity is essential (McGrail et al., 2017; Schwartz, 2018; Sigfússon et al., 2018). Oilwell cement provides a hydraulic barrier between the formation and casing, preventing fluid migration and interzonal cross-flow, thereby ensuring safe and efficient fluid management within the wellbore (Ramadan et al., 2019; Su et al., 2023; Eytayo et al., 2024). A major threat arises from exposure to corrosive gases such as H<sub>2</sub>S, which may originate from microbial sulfate reduction, thermochemical processes, thermal degradation, or intentional injection (Yan et al., 2022; Bois et al., 2025). H<sub>2</sub>S chemically reacts with cement phases, altering the cement matrix, and can migrate

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through inherent microcracks and other defects in cement, substantially compromising wellbore integrity (Zhang et al., 2013a; Choubineh et al., 2019). H<sub>2</sub>S-induced alteration of oilwell cement can result in leakage, environmental contamination, and serious safety hazards, as depicted in Fig. 1.

Due to the highly toxic and corrosive nature of H<sub>2</sub>S, research on its effects on the alteration of oilwell cement is limited (Kutchko et al., 2007; Zhang et al., 2013b; Li et al., 2015a). Previous studies on the alteration effects of H<sub>2</sub>S on oilwell cement have focused on various aspects, including the evolution of the cement's chemical composition, microstructural changes, degradation of mechanical properties, and the development of preventive measures (Yin et al., 2024). H<sub>2</sub>S can penetrate the pores and move along the microcracks in oilwell cement, leading to the dissolution of cement hydration products such as calcium silicate hydrate (C-S-H) and calcium hydroxide (Newell and Carey, 2013). The metal ions from dissolution of hydration products, particularly iron and calcium ions, subsequently react with H<sub>2</sub>S to form secondary minerals like pyrite and gypsum (Andreani et al., 2009). Kutchko et al. (2015) proposed that another significant product of the reaction between H<sub>2</sub>S and cement is ettringite, which is generated under the effects of “sulfate attack”, with sulfate originating from the Fe(III)-induced oxidation of H<sub>2</sub>S. These secondary minerals, characterized by poor bonding properties and expansive behaviors, accumulate within the cement matrix, resulting in the formation of microcracks and large pores (Peng et al., 2022). Consequently, this leads to reduced cement strength, increased permeability, and compromised structural stability (Glasser, 1996; Hawthorne et al., 2011; Um et al., 2012). Incorporating additives into cement is a widely adopted strategy for mitigating H<sub>2</sub>S-induced cement alteration (Idriss et al., 2001; Liu et al., 2021). The incorporation of additives into cement aims to accomplish three principal goals: reducing permeability, improving chemical stability, and creating physical barriers. Research has demonstrated that materials including nano-silica, nano-clay, asphalt, fibers, and fuzzy-ball fluid loss control additives effectively reduce cement permeability, thus decelerating the alteration process (Lecolier et al., 2010; Mei et al., 2022; Okere et al., 2023). Additives such as silica fume, slag, and fly ash contribute not only to pore filling

but also to the reduction of the more susceptible calcium hydroxide, thereby enhancing overall stability (Kutchko et al., 2015). Additionally, the integration of elastic particles like rubber or latex polymers can establish flexible barriers within the cement matrix (Wang et al., 2014). Certain metal compounds undergo alteration in preference to cement hydration products, forming stable compounds and demonstrating a “sacrificial protection” mechanism (Teng et al., 2017; Xu et al., 2018b). Building upon the alteration-inhibiting mechanisms of these materials, ongoing research continues to yield novel protective strategies (Larki et al., 2019; Bois et al., 2025).

The alteration of oilwell cement can be characterized using various technological methods, including CT scanning, SEM, EDS, FTIR, among others (Lecolier et al., 2010; Beck et al., 2016). CT scanning technology is a reliable technique for studying the alteration of porous media (Gan et al., 2020). It allows for the reconstruction of the spatial structure of materials and the calculation of their parameters (Kashim et al., 2019). CT scanning provides an accurate analytical tool for analyzing the alteration processes and microstructural evolution of cement. In the context of interactions between cement-based materials and acidic gases, CT scanning has also been introduced and applied (Um et al., 2012; Andrew et al., 2014; Garing et al., 2017). With high-precision CT scanning systems, the evolution of pore structures during the cement-acid gas reaction can be visually observed (Wang et al., 2013). Applying CT scanning to investigate H<sub>2</sub>S-induced alterations along fractures is expected to be efficient. However, it should be noted that CT scanning cannot distinguish minerals with very similar X-ray attenuation values (Gan et al., 2020). Therefore, combining this technique with other methods, such as SEM and FTIR, will facilitate a more precise analysis of the samples (Lebedev et al., 2017). Additionally, most previous studies have focused on static alteration conditions (Kutchko et al., 2011; Um et al., 2012). In the presence of microcracks within the cement, H<sub>2</sub>S can induce dynamic alteration of the cement matrix (Andreani et al., 2009).

To date, CT visualization of the alteration process induced by H<sub>2</sub>S flow along a leaking channel in oilwell cement, which causes more intense alteration than static conditions and better mimics actual wellbore leaking scenarios, has not been reported. First, the

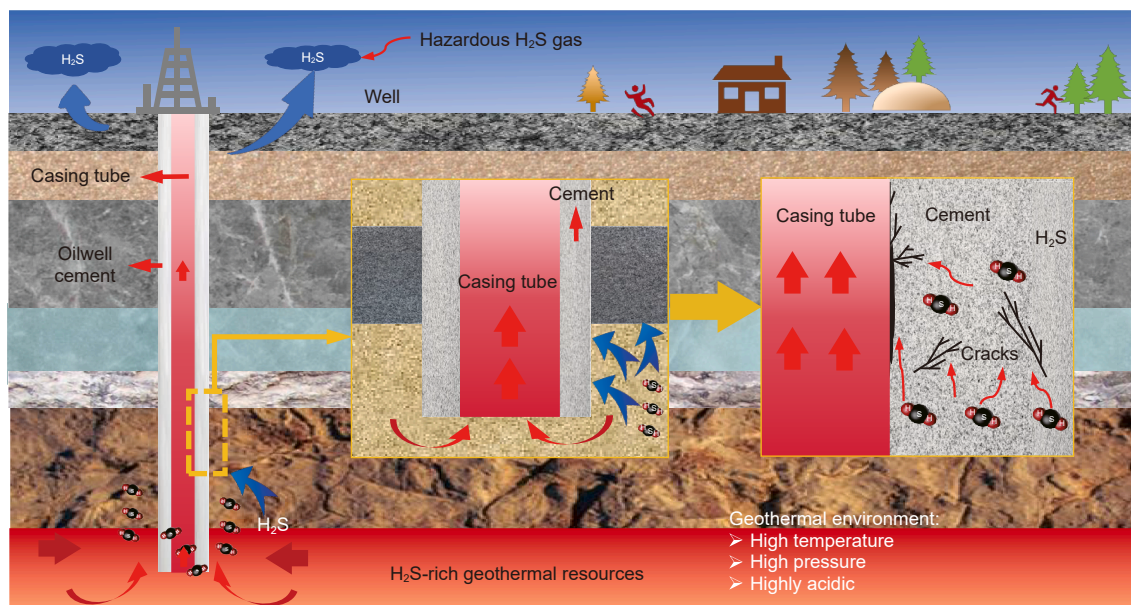


Fig. 1. Leakage in oilwell cement caused by acidic gas alteration.

alteration patterns, morphological features, and underlying chemical and microstructural mechanisms remain poorly understood. Second, the evolution of permeability in the cement matrix due to such flow-driven alteration has not been quantitatively investigated. Furthermore, the effect of high temperature (130 °C or above) on H<sub>2</sub>S-induced alteration of oilwell cement has not been adequately investigated in previous studies.

Given these issues, this study aims to investigate the alteration of oilwell cement resulting from H<sub>2</sub>S flow along a leaking channel under high-temperature and high-pressure condition. CT scanning was used to assess the alteration patterns and morphological features changes caused by H<sub>2</sub>S flow. Subsequently, microstructural evolution induced by H<sub>2</sub>S migration was examined using characterization techniques including SEM-EDS and FTIR. Finally, the mechanisms governing H<sub>2</sub>S-induced cement alteration under high temperature were elucidated. The findings are expected to provide critical insights for assessing the long-term integrity of wellbore systems in geothermal energy extraction operations.

## 2. Materials and methods

### 2.1. Preparation of cement samples

The experiment was conducted using Class G oilwell cement. The composition of the Class G oilwell cement, as determined by X-ray fluorescence (XRF) analysis, is presented in Table 1. A mixture consisting of Class G oilwell cement, quartz sand, silica fume, a dispersant, a water-reducing agent, a defoamer, and water was prepared and thoroughly mixed using a high-speed stirrer. The quartz sand had a fineness of 70 mesh and was added at a dosage of 8% by weight. SiO<sub>2</sub> is the primary component of the quartz sand used in the experiment, with a content exceeding 99%. The silica fume utilized in this study was obtained from a power plant and exhibited an average particle diameter of 0.2 μm and a specific surface area of 24 m<sup>2</sup>/g. It was incorporated into the mixture at a concentration of 10% by weight. The chemical composition of the silica fume consisted of more than 96% SiO<sub>2</sub>, with impurities comprising metal oxides-including sodium oxide, calcium oxide, iron oxide, and aluminum oxide-as well as trace amounts of elemental carbon and silicon. The total impurity content accounted for less than 4% of the total mass. The incorporation of quartz sand and silica fume into oilwell cement is designed to enhance its overall performance, particularly in terms of high-temperature resistance and alteration durability (Idriss et al., 2001; Liu et al., 2021). The dispersant (G33S) was included at 3% and comprised AMPS, low-molecular weight amides, and polymer-modified polyhydroxy carboxylic acid, while the water-reducing agent (SXY-2) was added at 1.5%, being a powdered ketone-aldehyde polymer that enhances the flowability of the cement slurry. To prevent bubble formation during mixing, the defoamer was introduced at a concentration of 0.05%. The water-to-solid ratio for this experiment was set at 0.44.

The mixed paste was poured into a mold with a diameter of 40 mm and a height of 30 mm, which included a 5 mm diameter flow channel in the middle to simulate H<sub>2</sub>S seepage in pre-existing defects within oilwell cement. The prepared cement samples were cured in water at 1 atm and 62 °C for 28 days to promote the hydration process. After curing, the samples were removed and dried for subsequent use.

**Table 1**  
XRF test result of the Class G oilwell cement.

Composition	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	MgO	CaO	K <sub>2</sub> O	Na <sub>2</sub> O	SO <sub>3</sub>
Amount, %	22.31	3.60	4.76	1.29	64.06	0.41	2.48	0.96

### 2.2. H<sub>2</sub>S flow-oilwell cement alteration test

The cured cement samples were placed into the reaction vessel. Photographs of the cured cement sample and the reaction vessel are presented in Fig. 2(a) and (b), respectively. The cement sample was securely fixed against the wall of the vessel using silk threads that wrapped around its exterior, ensuring that the H<sub>2</sub>S-containing liquid only flowed through the channel in the middle. The temperature for the H<sub>2</sub>S reaction was set at 130 °C, with partial pressures of H<sub>2</sub>S and N<sub>2</sub> maintained at 1 and 9 MPa, respectively. This pressure configuration was designed to mimic the conditions of high-sulfur gas reservoirs in China's Sichuan Basin (Huang et al., 2010). Although such reservoirs typically contain significant CO<sub>2</sub> rather than N<sub>2</sub>, the present study focuses on elucidating the specific mechanisms of H<sub>2</sub>S interaction with oilwell cement. The inclusion of CO<sub>2</sub> was unnecessary and risked obscuring H<sub>2</sub>S-specific effects. Therefore, in line with the approaches of Li et al. (2015b), Cheng et al. (2016) and Mei et al. (2019), inert N<sub>2</sub> was used to replace all non-H<sub>2</sub>S gases to ensure experimental controllability.

The cement samples were allowed to react with H<sub>2</sub>S for a duration of 14 days, during which continuous monitoring was performed to maintain constant gas pressure. Although the duration of the alteration tests was shorter than the actual service life of a wellbore in engineering applications, the results provide valuable reference data for inferring long-term corrosion behavior. First, short-term tests help clarify the mechanisms and characteristics of cement alteration, including the types and morphology of changes, as well as the chemical reactions between cement and H<sub>2</sub>S (Li et al., 2015a; Cheng et al., 2016). Furthermore, such tests yield data relevant to long-term changes in actual engineering, which can be used to develop alteration models and supply necessary parameters for numerical simulations. Additionally, short-term tests serve as a comparative benchmark for evaluating different materials or treatment methods, enabling researchers to quickly identify the most effective materials or anti-alteration measures (Xu et al., 2018a; Mei et al., 2019). In previous studies, an alteration duration of 14 days has been commonly adopted, as seen in the work of Mei et al. (2019) and Yin et al. (2024).

### 2.3. Test methods

#### 2.3.1. CT scanning test

CT scanning was conducted on both the unaltered cement sample (i.e., those not exposed to H<sub>2</sub>S) and the cement sample that had been altered by H<sub>2</sub>S solution for 14 days. This was done to investigate the structural evolution of oilwell cement under H<sub>2</sub>S alteration, particularly focusing on pore distribution, porosity, and the morphology of pore structures. CT scanning is a non-invasive and non-destructive testing technique that utilizes three-dimensional imaging (Miao et al., 2020). It determines the spatial distribution of X-ray attenuation coefficients of an object by measuring its X-ray projections from multiple angles, thereby allowing for the representation of variations in material structure. The analysis of CT images is based on measuring the X-ray attenuation properties of different materials, with results influenced by their atomic numbers and densities (Gan et al., 2024).

In this experiment, a Zeiss Xradia 410X X-ray CT scanner was utilized for CT scanning. The operational parameters of the CT scanner were as follows: 80 kV voltage, 125 μA current, 10 W power, and a voxel size of 11.08 μm. The scanning process involved rotating the samples 360° around the Z axis of the workbench in predetermined increments to obtain projection image signals from various angles. Changes in the grayscale values of the CT images were analyzed to identify differences in material density and porosity before and after H<sub>2</sub>S alteration. The processing of CT

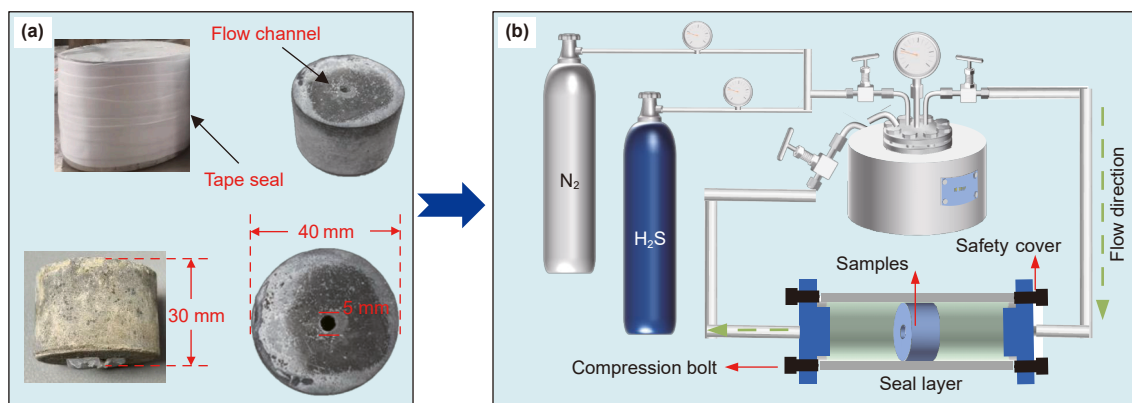


Fig. 2. (a) Dimensions of the cement sample; (b) schematic diagram of the oilwell cement alteration testing.

images consisted of data import, denoising filtration, and segmentation, all performed using Avizo software. Segmentation is a vital step in CT data processing, wherein each voxel is categorized into discrete phases corresponding to different material constituents. Histogram-based thresholding, assigning each phase to a specific interval within the grayscale histogram are employed. The segmentation methodology adopted in this study aligns with established approaches reported in prior literature, wherein pores are differentiated from the cement matrix according to variations in grayscale intensity (Wan et al., 2014; Miao et al., 2020; Wang et al., 2021). The segmentation threshold was identified by locating the valley between the first two peaks in the grayscale histogram. The CT scanning system and the method for slicing the samples are illustrated in Fig. 3. Specifically, slices P1 and P4 were taken 1 mm from the top and bottom surfaces of the sample, respectively, while slices P2 and P3 were evenly spaced in the middle of the sample.

### 2.3.2. Permeability test

Cement samples without artificial channels were selected for permeability testing, so as to understand how H<sub>2</sub>S alteration affects the permeability of the oilwell cement. The permeability of the samples was measured using a self-developed pulse permeability testing apparatus. This pulse testing device calculates the permeability coefficient of the sample by establishing the decay curve of the pressure difference across the upstream and downstream of the tested sample over time. The specific method for calculating the permeability coefficient follows the reference (Sun et al., 2018). In this experiment, the effective confining pressure was set at 3.5 MPa, and the pressure difference was set at 0.1 MPa.

### 2.3.3. SEM-EDS and FTIR tests

CT scanning does have limitations, such as challenges in visualizing materials at the microscopic level and characterizing chemical evolution processes. To address these limitations, CT scanning is supplemented by other advanced techniques such as SEM, EDS, and FTIR to achieve a comprehensive analysis of microstructural evolution of oilwell cement.

Small pieces of oilwell cement samples, both before and after H<sub>2</sub>S alteration, were analyzed using SEM and EDS. For SEM analysis, the cement samples were mounted on the observation stage, with observations conducted at magnifications of 100X and 2000X. EDS line scanning was performed from the wall of the flow channel toward the interior of the sample to monitor the weight percentage evolution of elements such as calcium, silicon, and sulfur.

Additionally, some cement samples were finely grounded, and the resultant powders were sieved through a screen with an aperture of 0.075 mm to prepare samples suitable for FTIR phase analysis. The obtained sample powders underwent FTIR testing to analyze and characterize the chemical features of the oilwell cement before and after H<sub>2</sub>S alteration.

## 3. Results

### 3.1. CT scanning results

#### 3.1.1. Demonstration of CT scanning slices perpendicular to the channel

Fig. 4 illustrates the CT scanning slices of the cement with a leakage channel, captured before and after exposure to H<sub>2</sub>S. These images depict the morphology of the cement at various locations before and after H<sub>2</sub>S alteration. As the H<sub>2</sub>S solution flows through the central channel of the cement, complex cracking patterns emerge near the inlet (as shown in C-P1 in Fig. 4), indicating significant damage to the structural integrity of the cement. The generation of expansive ettringite may lead to the cracking of the cement matrix (Kutchko et al., 2015). The secondary minerals formed through sulfide reactions exhibit weak bonding characteristics with the cement matrix, and under the dynamic flow of the H<sub>2</sub>S solution, these minerals may be washed away (Zhang et al., 2014). The changes observed from slices P1 to P4 demonstrate that the alteration effects diminish with increasing distance from the inlet along the flow direction of H<sub>2</sub>S. The morphological images indicate that there are no discernible macrocracks in the cement near the outlet (C-P4).

Fig. 5 presents slices illustrating pore distribution extracted from cement sample near the inlet and outlet both before and after the reaction with H<sub>2</sub>S. The images have undergone threshold segmentation to enhance the visibility of the pores, which are represented in blue. The porosity of the cement sample prior to exposure to H<sub>2</sub>S is significantly lower than that observed after the reaction. Furthermore, the interaction between H<sub>2</sub>S and the cement results in a marked increase in the formation of larger pores. This indicates that the reaction not only heightens the internal porosity of the cement but also enlarges the pore sizes. In regions adjacent to the H<sub>2</sub>S outlet (Fig. 5(b), C-bottom), small pores can be observed developing near the channel, a phenomenon that is absent in the unaltered sample. In comparison to Figs. 4 and 5 clearly illustrates the accumulation of microcracks near the inlet. These cracks extend from the channel into the cement matrix, displaying variability in their propagation direction; some extend perpendicular to the channel, while others run parallel. Notably,

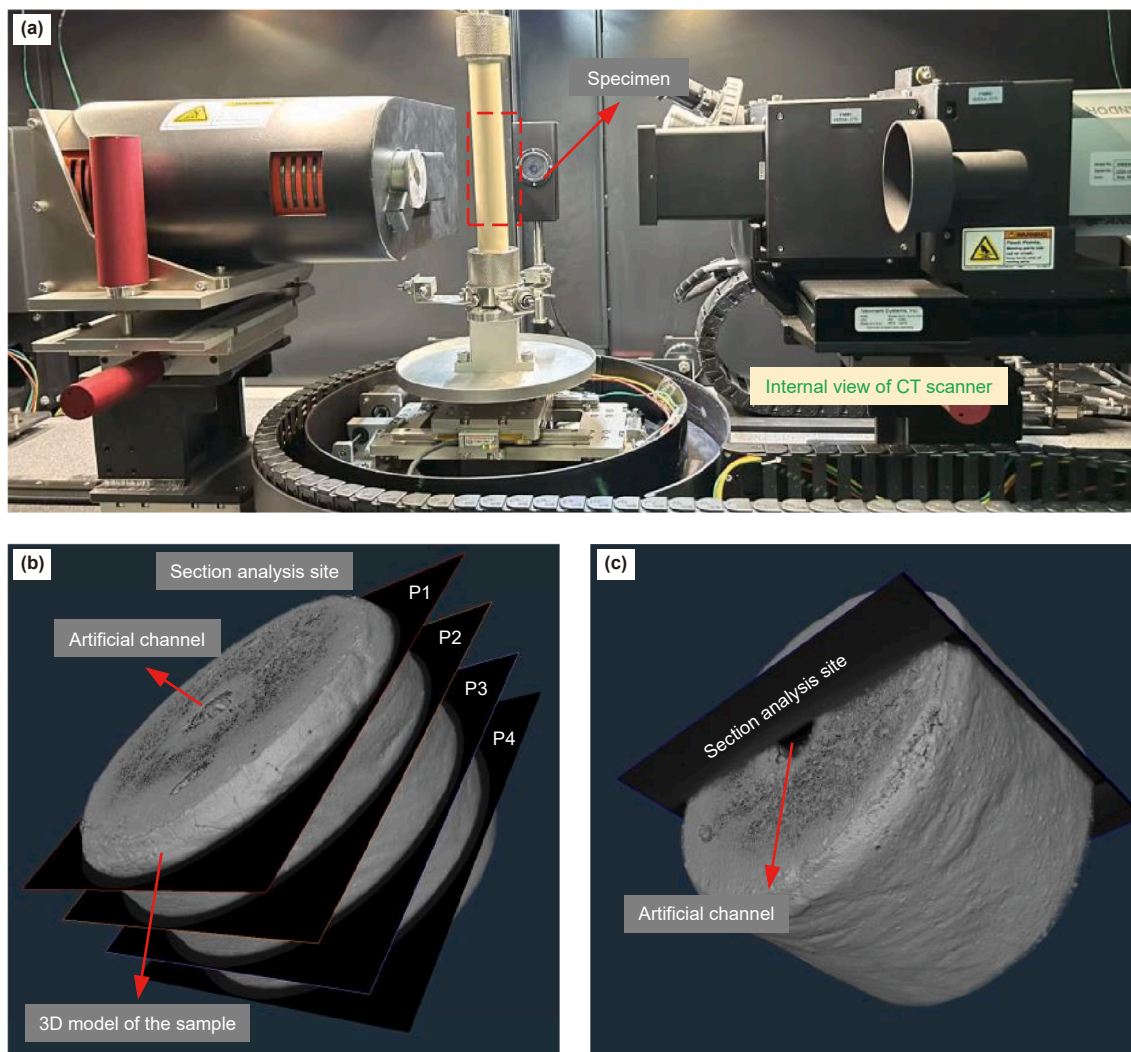


Fig. 3. (a) CT scanning testing system, (b) sample slices perpendicular to the channel, (c) sample slice along the channel.

certain microcracks are continuous, connecting larger pores. The alteration effect of  $H_2S$  causes the pore sizes within the cement to expand, thereby creating new weak points. In C-Top of Fig. 5, due to the proximity to the inlet, the effect of  $H_2S$  is intensified, resulting in a continual increase in secondary material formation, which exacerbates both dissolution and localized expansion forces. Consequently, this leads to the development of microcracks along the weak points of the pores. The corrosive impact of  $H_2S$  on cement mirrors the effects observed with the flow of  $CO_2$  solutions (Gan et al., 2022).

To analyze the distribution of porosity perpendicular to the channel, the area ratio of pores to the total plane area (excluding the channel) was extracted and is presented in Fig. 6. Significant changes in the porosity of the cement sample are observed before and after the reaction with  $H_2S$ . Prior to the reaction, the porosity distribution of the cement sample is relatively uniform, with most planes exhibiting porosity in the range of 2%–2.5%. Following the reaction with  $H_2S$ , the porosity in the majority of the cement sample has markedly increased, with the maximum porosity observed near the middle of the cement, reaching up to 4.817%. The porosity at the bottom of the sample is the lowest, approaching that of the unaltered sample, while the porosity near the inlet is approximately 3.5%. This

value is lower than that in the middle of the cement but higher than at the bottom. This phenomenon can be attributed to the relatively high concentration of corrosive species near the inlet region, which enhances dissolution-precipitation reactions within the cement. Based on conclusions drawn from similar experiments by Yin et al. (2024) this dissolution-precipitation mechanism primarily involves the dissolution of calcium hydroxide and ettringite, along with the precipitation of  $FeS$ ,  $FeS_2$ , and  $CaSO_4$ . Intensive dissolution and reprecipitation in this zone result in volumetric expansion and microstructural deformation of the cement matrix. The mid-channel region is characterized predominantly by dissolution, leading to gradual consumption of  $H_2S$ . As a consequence, the concentration of  $H_2S$  decreases toward the bottom of the channel, shifting the dominant reaction mechanism to precipitation and ultimately yielding reduced porosity. It is important to note that the resolution limit of CT scanning is 0.9 mm, and many small pores within the cement have diameters smaller than this threshold, rendering them unidentifiable (Gan et al., 2020). Consequently, the porosity obtained from CT scanning is lower than the actual porosity (Miao et al., 2020). Nonetheless, it remains feasible to utilize the CT scanning results to analyze relative changes in pore characteristics (Wang et al., 2021).

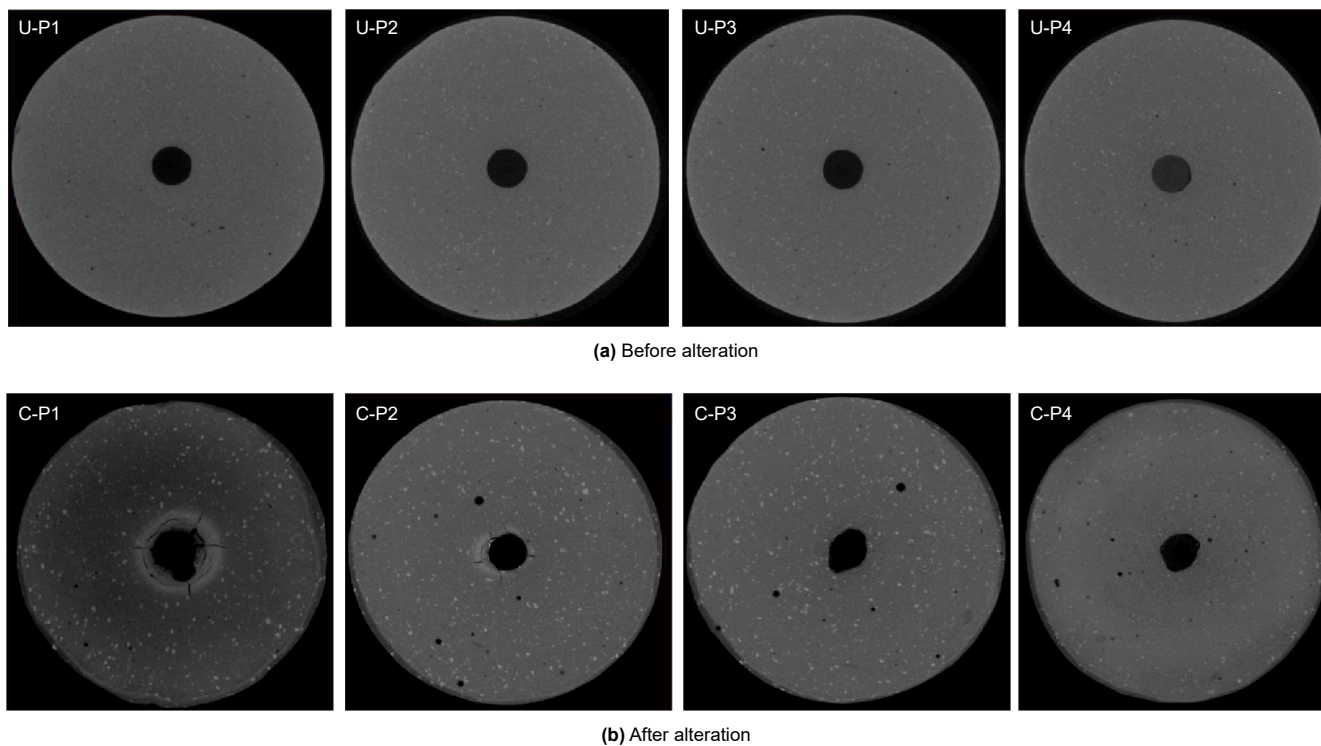


Fig. 4. Slices perpendicular to the channel: (a) before alteration; (b) after alteration (“C” refers to the sample after reaction with H<sub>2</sub>S, and “U” refers to the sample before reaction with H<sub>2</sub>S).

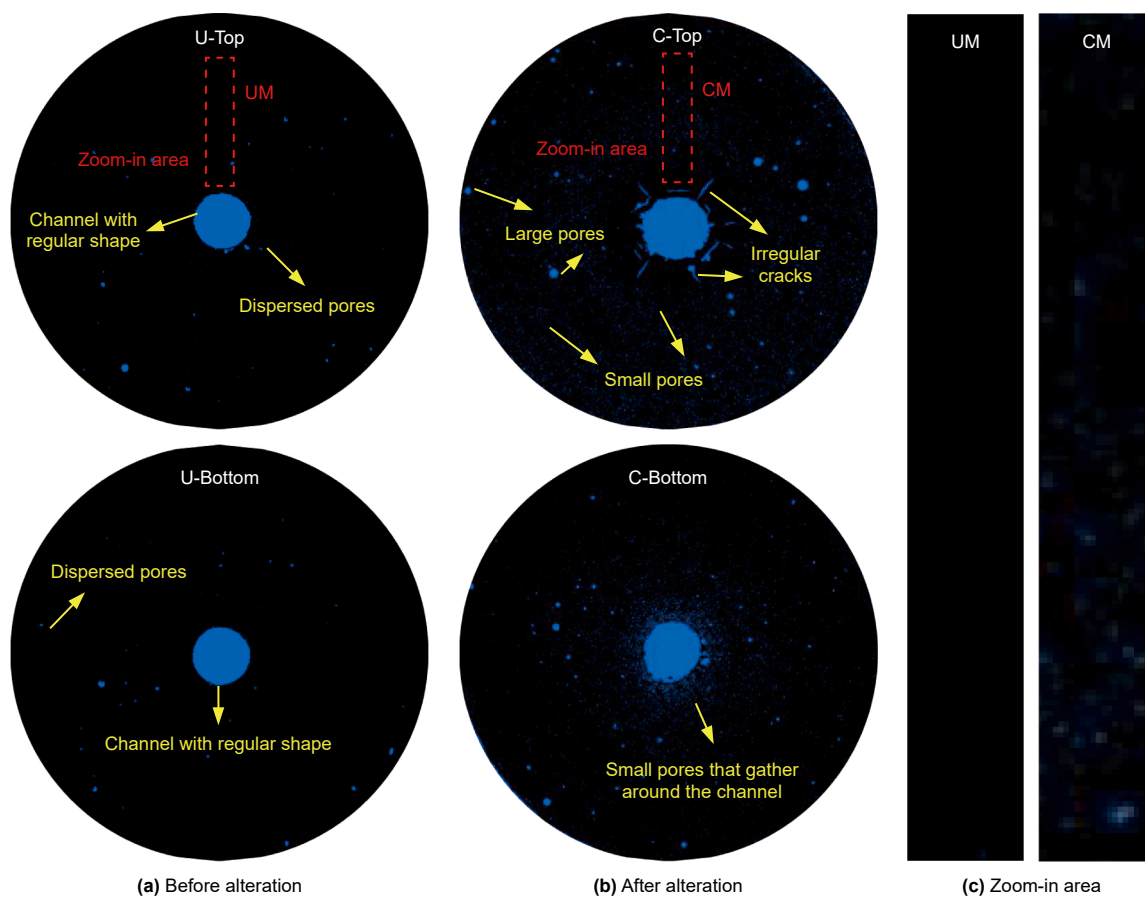


Fig. 5. Pore distribution of slices perpendicular to the channel: (a) before alteration; (b) after alteration; (c) zoom-in area (“C” refers to the sample after reaction with H<sub>2</sub>S, and “U” refers to the sample before reaction with H<sub>2</sub>S).

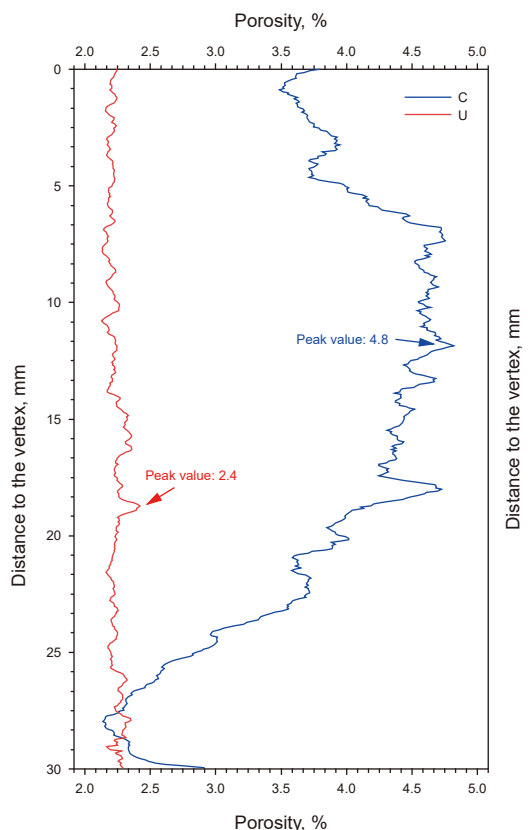


Fig. 6. Pore distribution of slices perpendicular to the channel (“C” refers to the sample after reaction with H<sub>2</sub>S, and “U” refers to the sample before reaction with H<sub>2</sub>S).

### 3.1.2. Slices along the channel

H<sub>2</sub>S permeates through the channel within the cement, leading to alteration to the cement matrix surrounding the channel. Consequently, it is essential to examine the slices along the channel. Fig. 7 presents morphological images of cement sample slices along the channel before and after exposure to H<sub>2</sub>S. After exposure to H<sub>2</sub>S, the cement exhibits increased crack propagation and a greater number of larger pores. The cracks situated near the inlet appear to be approximately parallel to the channel within the central region of the cement. The high gray value areas, highlighted in white in the figure, correspond to quartz sand. A comparison of the quartz sand indicates that the gray values of the cement matrix prior to the reaction with H<sub>2</sub>S are higher than those observed after the reaction. Some studies suggest that the decrease in gray value of the cement following exposure to acidic

media is attributable to an increase in fine pore structures resulting from the reaction (Li et al., 2015b, Miao et al., 2020).

The slices depicted in the previous figure were processed using threshold segmentation to clearly distinguish the pores (depicted in blue) from the matrix (shown in black), as illustrated in Fig. 8. The unaltered cement samples contain fewer pores, which are also smaller in size, and the channel wall surfaces appear relatively smooth. Following alteration by H<sub>2</sub>S, there is a notable increase in the overall porosity of the cement sample, characterized by a greater number of larger pores and the emergence of microcracks. The newly formed pores are primarily situated near the channel, while some microcracks extend inward along the channel, with internal microcracks accumulating in regions where pore expansion occurs. The distribution of these cracks in areas adjacent to the alteration indicates that these regions have experienced significant degradation. The alteration of cement by acidic gases leads to varying degrees of modification across different regions (Li et al., 2015b). Furthermore, areas with a high concentration of cracks exhibit a marked increase in the roughness of the cement wall surfaces. It can be reasonably inferred that this change is a consequence of the chemical reactions between H<sub>2</sub>S and the chemical composition of the cement, resulting in the dissolution of some primary minerals.

The analysis of Figs. 4–8 indicates that H<sub>2</sub>S not only induces significant alteration to the porosity characteristics of the cement matrix but also causes deformation of the channel. Consequently, the entire channel has been segmented, and the ratio of the channel area to the corresponding plane area has been calculated, as illustrated in Fig. 9. In the unaltered cement sample, the flow channel in the central region maintains a generally uniform shape and a smooth surface. However, following exposure to H<sub>2</sub>S, the wall roughness of the channel in the middle of the cement increases, resulting in an irregular shape. Near the inlet, the dimensions of the channel expand due to the erosive action of H<sub>2</sub>S. In regions adjacent to the inlet, some of the channel areas approximate those of the unaltered sample, which can be attributed to the increase in sample volume after alteration. Significant alteration is also observed in the middle of the channel, leading to an expansion of the channel volume. However, as moves closer to the outlet, this expansion effect gradually decreases. Non-uniform variations in channel volume, attributable to the synergistic effects of chemical etching and secondary crystal growth, have been similarly reported by Gan et al. (2020).

### 3.2. Permeability of cement matrix

Under the corrosive influence of H<sub>2</sub>S, the porosity of oilwell cement increases, subsequently leading to enhanced permeability of the cement matrix. Cylindrical cores were extracted from

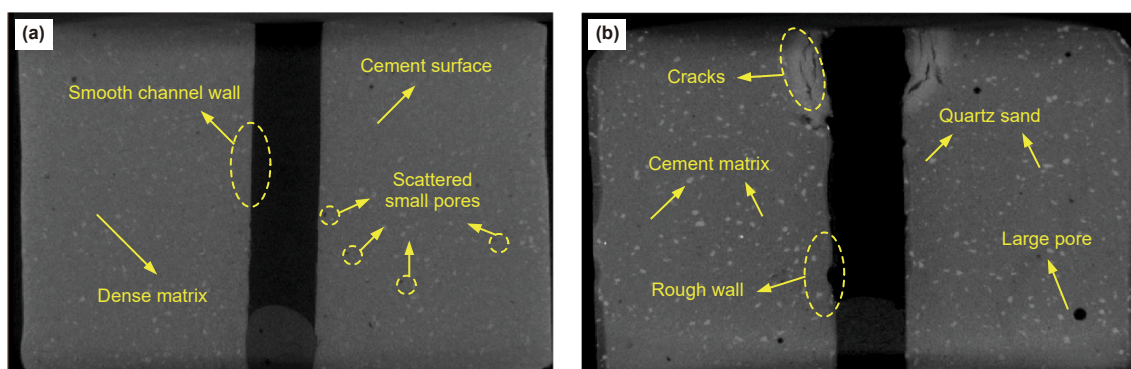


Fig. 7. Slices along the channel: (a) before alteration; (b) after alteration.

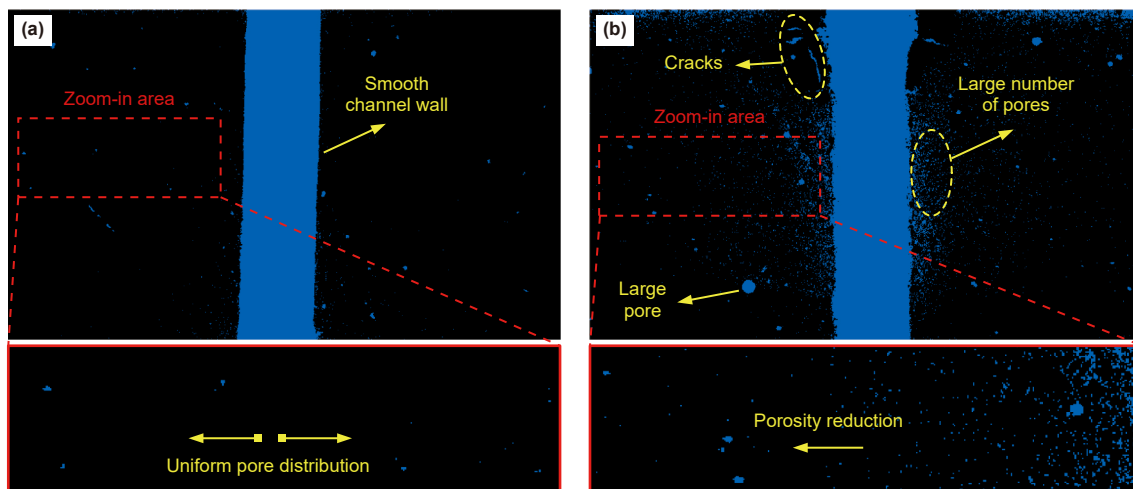


Fig. 8. Pore distribution of slices along the channel: (a) before alteration; (b) after alteration.

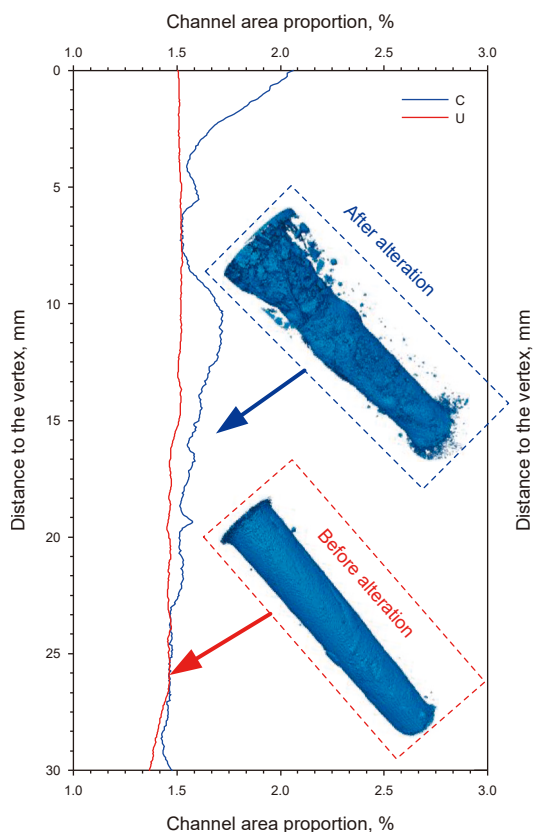


Fig. 9. Changes in channel dimensions and comparison of 3D structures (“C” refers to the sample after reaction with H<sub>2</sub>S, and “U” refers to the sample before reaction with H<sub>2</sub>S).

cement sample that had undergone alteration by H<sub>2</sub>S. These samples, along with unaltered cement samples of equivalent dimensions, were subjected to permeability testing. Notably, the samples used for permeability testing did not contain any channel; therefore, the reported permeability values reflect the permeability of the cement matrix. For each category, three samples were tested, one set consisting of cement samples altered by H<sub>2</sub>S for 14 days, and another set comprising unaltered cement samples. This approach was adopted to yield average permeability values

Table 2  
Permeability of cement matrix.

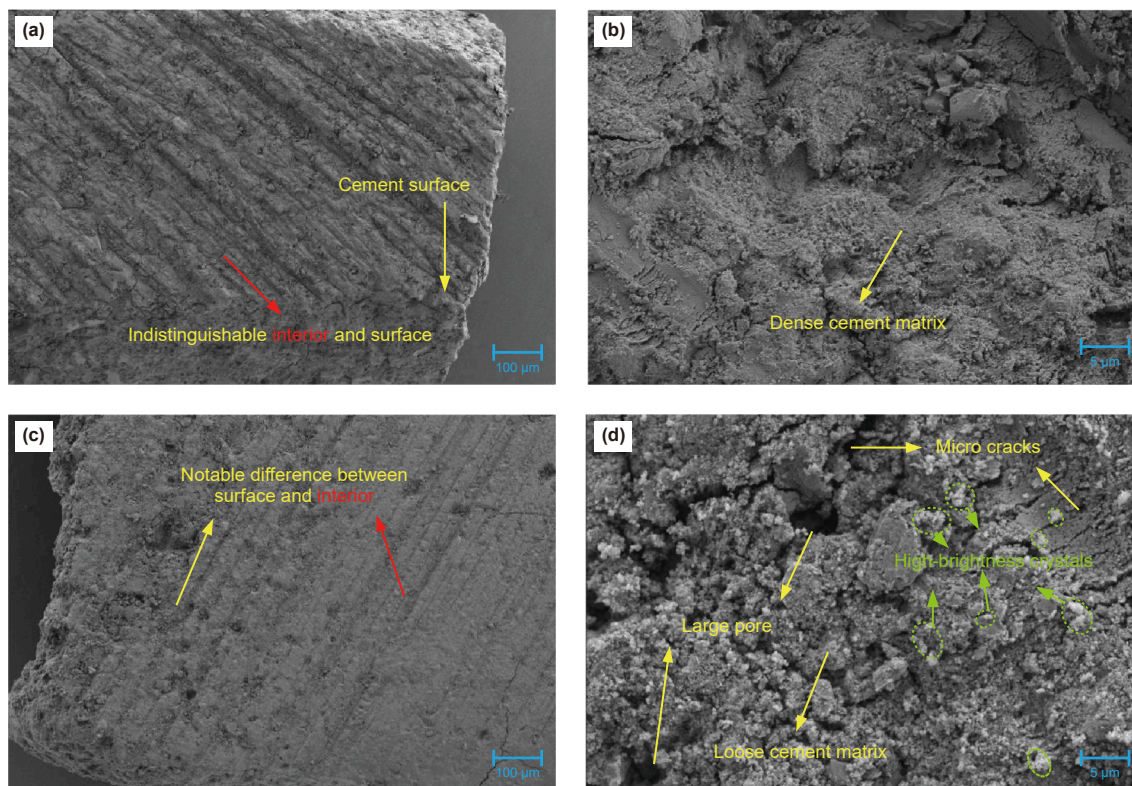
Sample name	Permeability, mD	
	Value	Average value
U-#1	0.064	0.060
U-#2	0.059	
U-#3	0.056	
C-#1	0.091	0.087
C-#2	0.086	
C-#3	0.083	

and minimize error. The results of the permeability tests are presented in Table 2. The permeability values of the unaltered cement samples were recorded as 0.064, 0.059, and 0.056 mD, yielding an average value of 0.060 mD. In contrast, the permeability values of the cement samples altered by H<sub>2</sub>S for 14 days were 0.091, 0.086, and 0.083 mD, resulting in an average value of 0.087 mD. Compared to the unaltered samples, the average permeability of the cement samples altered by H<sub>2</sub>S for 14 days increased by 45%. Unlike CO<sub>2</sub>, which may cause pore blockage and subsequently reduce the permeability of cement, H<sub>2</sub>S markedly enhances the permeability of cement (Xu et al., 2018a).

### 3.3. Microstructure and chemical test results

#### 3.3.1. SEM test results

CT scanning results illustrate the overall structural changes in the cement; however, they fall short in displaying the morphological changes at specific locations within the material. To address this limitation, SEM experiments were conducted to examine the microstructural differences in the cement caused by H<sub>2</sub>S exposure. Fig. 10 presents SEM images of the cement samples before and after the reaction with H<sub>2</sub>S. As shown in Fig. 10(a), the unaltered cement exhibits a uniform texture characterized by low porosity and a minimal number of large pores. When magnified 2000 times, as depicted in Fig. 10(b), the cement particles appear fine, exhibiting few pores with no significant voids present. The structural integrity of the cement sample is well preserved, with virtually no microcracks observed. Conversely, the cement following the reaction with H<sub>2</sub>S, as illustrated in Fig. 10(c), reveals the formation of larger pores. A comparison between the edges and the interior of the sample indicates that the porosity and the abundance of large pores at the edges are significantly lower than those found in the interior. When



**Fig. 10.** SEM images of the oilwell cement: (a) before alteration, magnification 100X; (b) before alteration, magnification 2000X; (c) after alteration, magnification 100X; (d) after alteration, magnification 2000X.

localized areas are examined at the same magnification (2000 times), as seen in Fig. 10(d), the microstructure of the cement post-alteration appears markedly different from that prior to alteration. The altered cement exhibits larger pores, cracks, and a transformation of the cement material into unevenly fragmented larger particles. This microstructural transformation is typical of cement altered by acid gases (Li et al., 2015b). Additionally, the conspicuous formation of new, high-brightness crystals, likely identified as FeS and FeS<sub>2</sub> based on the study by Yin et al. (2024) indicates that the reaction between H<sub>2</sub>S and cement follows a dissolution-precipitation mechanism rather than adsorption. This further suggests that sulfur is incorporated into the crystal lattice rather than being surface-adsorbed. The structural integrity of the cement is significantly compromised after the reaction with H<sub>2</sub>S. These findings underscore the pronounced impact of H<sub>2</sub>S on the physical and chemical stability of the cement. The mineral alterations that occur in the aftermath of H<sub>2</sub>S exposure not only modify the cement's microstructure but also adversely affect its sealing capacity and durability under prolonged service conditions.

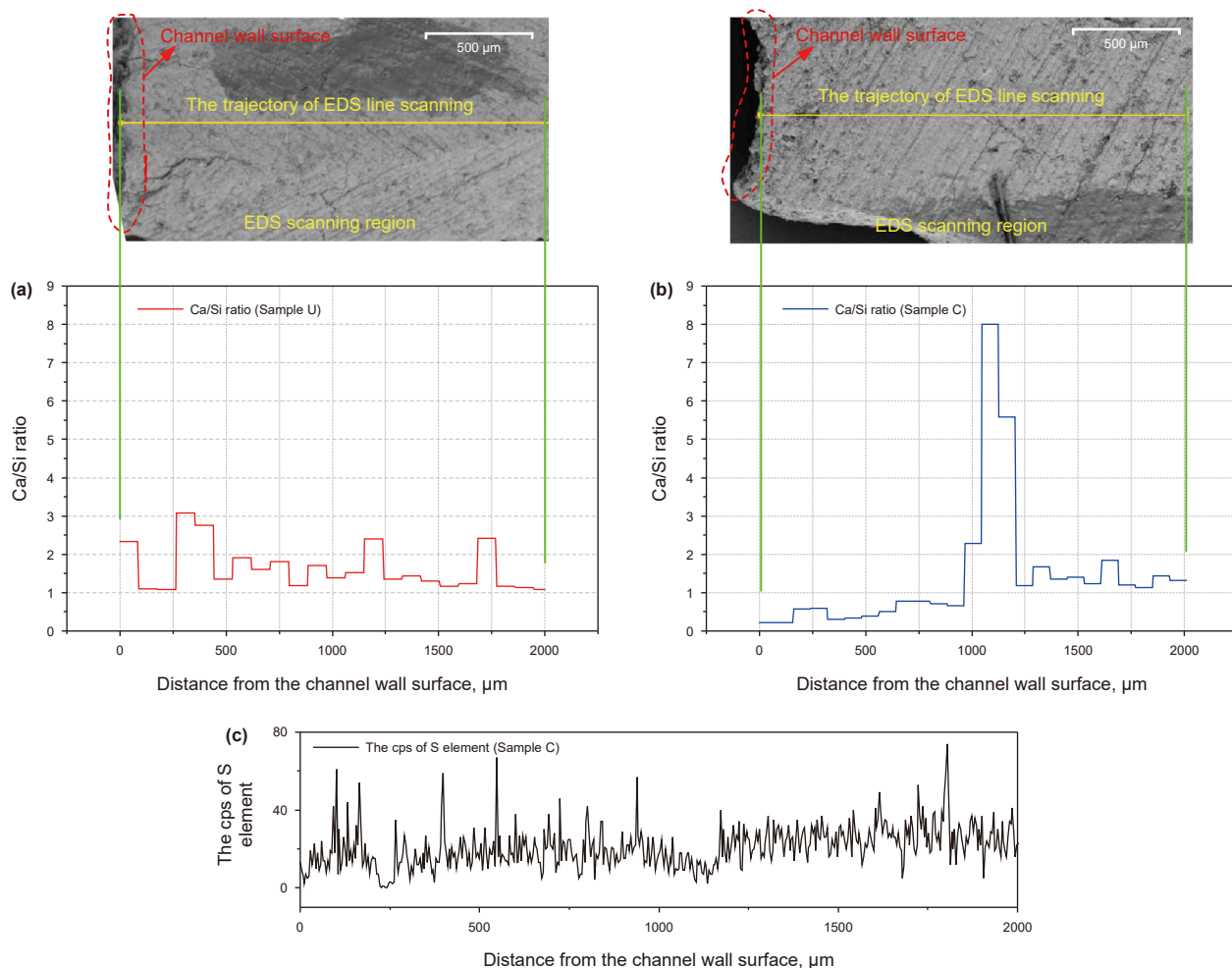
### 3.3.2. EDS test results

EDS element scanning provides a visual representation of the elemental distribution changes within the cement. As depicted in Fig. 11(a), the Ca/Si ratio in unaltered cement is relatively similar. When the EDS scanning is conducted from the channel wall surface towards the interior, the Ca/Si ratio ranges approximately from 1 to 3. In contrast, following exposure to H<sub>2</sub>S, the Ca/Si ratio in the cement decreases, as shown in Fig. 11(b). EDS line scanning from the interior to the channel wall surface reveals a gradual decline in this ratio. This reduction indicates that calcium ions are leaching from the cement matrix. Previous studies have established that the interaction between H<sub>2</sub>S and cement facilitates the

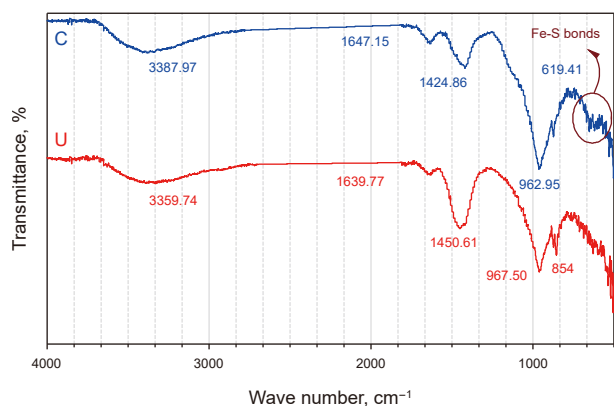
precipitation of metal ions (Zhang et al., 2013a; Kutchko et al., 2015). These metal ions can react with sulfide ions or sulfate ions produced from the dissolution of H<sub>2</sub>S, resulting in the formation of new substances (Yin et al., 2024). Notably, the estimated thickness of the decalcification zone on the cement surface is approximately 1 mm. The phenomenon of acid gases leading to delamination in cement structures has been noted, and additional findings reinforce this observation (Kutchko et al., 2007). However, this does not suggest that the alteration depth of H<sub>2</sub>S is confined to 1 mm, as the EDS scan results for sulfur demonstrate that the stable distribution of S element extends further into the cement, as illustrated in Fig. 11(c). The increase in porosity further supports the depth of H<sub>2</sub>S-induced alteration. This phenomenon can be explained by the reaction of H<sub>2</sub>S with the cement upon contact, leading to the leaching of metallic ions that are continuously carried away, forming a decalcified layer near the channel wall. Subsequently, H<sub>2</sub>S penetrates the interior of the cement sample, causing the internal metallic ions to migrate toward the channel and resulting in the dissolution of the cement's native substances, ultimately increasing porosity.

### 3.3.3. FTIR test results

FTIR is highly sensitive to chemical bonds and can effectively analyze the material changes in cement samples before and after alteration by H<sub>2</sub>S. Fig. 12 presents a comparison of the FTIR results for unaltered cement samples with those altered by H<sub>2</sub>S for 14 days. Characteristic peaks are present in both samples at 3396, 1636, 1421, 961, 873, and 619 cm<sup>-1</sup>. The peak at 3396 cm<sup>-1</sup> is typically associated with the stretching vibration of O–H groups, indicative of calcium hydroxide's presence in the cement. The peak at 1636 cm<sup>-1</sup> likely pertains to the bending vibration of O–H in bound water. The peak at 1421 cm<sup>-1</sup> primarily corresponds to the



**Fig. 11.** EDS line scanning results of the oilwell cement: (a) Ca/Si ratio before alteration; (b) Ca/Si ratio after alteration; (c) the cps (counts per second) of sulfur (S) element (cps is used to quantify the concentration of elements within a sample) (“C” refers to the sample after reaction with H<sub>2</sub>S, and “U” refers to the sample before reaction with H<sub>2</sub>S).



**Fig. 12.** FTIR test results (“C” refers to the sample after reaction with H<sub>2</sub>S, and “U” refers to the sample before reaction with H<sub>2</sub>S).

stretching vibration of C–O. The peaks at 961 and 873 cm<sup>-1</sup> are characteristic absorption peaks typically linked to Si–O or S–O bonds, which may arise from sulfur-cement derivatives or sulfate compounds. The peak at 619 cm<sup>-1</sup>, positioned within the low-frequency range, is associated with the vibrational modes of Fe–S bonds. This observation aligns with findings from previous studies indicating the formation of metal sulfides as a result of the

interaction between H<sub>2</sub>S and cement (Jacquemet et al., 2012). Generally, oilwell cement does not produce sulfides following hydration (Taylor, 1965; Guner et al., 2017). However, sulfides generated from the alteration of cement by H<sub>2</sub>S, can significantly impact the mechanical properties and microstructure of the cement samples, leading to increased porosity and decreased strength (Jung and Um, 2013). Experiments conducted by Yin et al. (2024) under comparable conditions demonstrated that H<sub>2</sub>S induces the dissolution of calcium hydroxide and ettringite in oilwell cement, along with the formation of FeS, FeS<sub>2</sub>, and CaSO<sub>4</sub>. Notably, the formation of thaumasite, a sulfate attack product typically observed in cement under low-temperature conditions, was not detected, as this phase is thermodynamically unstable at high temperatures.

#### 4. Discussion

During the seepage alteration of H<sub>2</sub>S along the cement channel, the porosity on both sides of the channel is observed to increase, with a proliferation of larger pores. The alteration effects are more pronounced as proximity to the channel increases. Following H<sub>2</sub>S alteration, the channel walls exhibit roughening, and the sizes of the channel expand. The microstructural changes in cement, as evidenced by FTIR in this study and supported by earlier results from Yin et al. (2024) on H<sub>2</sub>S-cement interactions under flowing

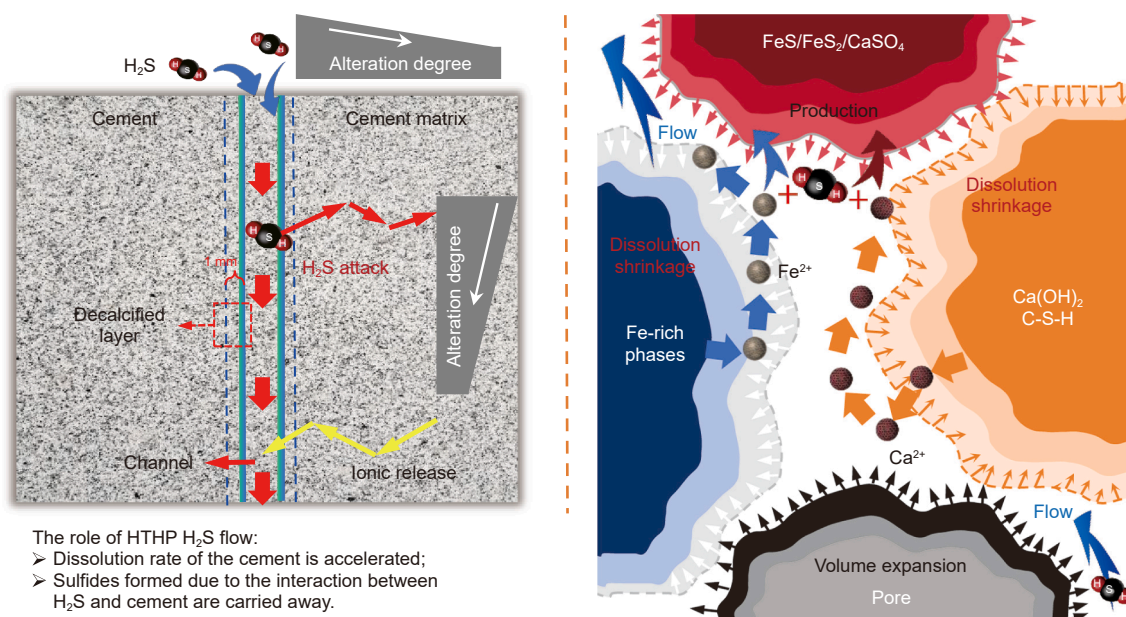


Fig. 13. Schematic representation of cement alteration induced by the flow of H<sub>2</sub>S along the channel.

conditions, are consistent with a chemical mechanism involving dissolution of ettringite and Ca(OH)<sub>2</sub>, coupled with precipitation of FeS, FeS<sub>2</sub>, and CaSO<sub>4</sub>. The incorporation of sulfur into the crystal lattice renders the deleterious effects of this chemical process persistent.

The flow alteration of cement by H<sub>2</sub>S in this study has a pronounced effect, which supplements previous results from one-phase alteration tests (Li et al., 2015a), immersion alteration tests (Zhang et al., 2013b), and dry gas alteration tests (Hawthorne et al., 2011). By comparing the dynamic alteration results of this study with prior static alteration findings, it was determined that flow-induced alteration by H<sub>2</sub>S results in more pronounced degradation of the cement (Ji and Zhu, 2013; Xu et al., 2018a). This phenomenon has also been observed in certain studies concerning the flow-induced alteration of CO<sub>2</sub> along channel within cement (Gan et al., 2022). This increased degradation may be attributed to the flowing acidic gas solution, which transports away ions leached from the cement, thus perpetuating the dissolution process of the cement. Conversely, in static reactions, a decrease in the reaction rate between acidic gases and cement has been noted, potentially due to the blockage of certain pores by secondary biomass (Jacquemet et al., 2012; Zhang et al., 2013a). The surfaces of the cement that come into direct contact with the H<sub>2</sub>S solution may become entirely decalcified, while the hydration products within the cement continue to dissolve, migrate toward the channels, and are ultimately removed. This leads to an increase in both the porosity and permeability of the cement matrix, accompanied by the development of structural layering. The flow-alteration process is illustrated and summarized in Fig. 13. As is known from previous studies, Ca(OH)<sub>2</sub> reacts rapidly with H<sub>2</sub>S and contributes most significantly to the early-stage evolution of the pore structure (Zhang et al., 2013b; Kutchko et al., 2015). In contrast, the degradation of C–S–H gel results in prolonged and fundamental weakening of the cement matrix, accompanied by continuous development of the pore network (Jacquemet et al., 2012). Furthermore, the dissolution of the cement results in a decline in the alteration resistance of the substrate (Yin et al., 2024). Therefore, the flow-induced alteration of cement by H<sub>2</sub>S warrants significant attention.

Previous studies on the alteration of oilwell cement by H<sub>2</sub>S have primarily focused on lower temperatures, such as 50, 90, and 100 °C, among others (Kutchko et al., 2011, 2015; Zhang et al., 2014). In the context of geothermal resource extraction, a temperature of 130 °C is classified as moderate-temperature geothermal. In environments with abundant H<sub>2</sub>S, the synergistic effects of high temperature, high pressure, and flow continuously facilitate the alteration of the cement (Zhang et al., 2023). This finding holds significant implications for high-temperature geothermal extraction, oil and gas field development.

## 5. Conclusions

An experiment was conducted to investigate the alteration of oilwell cement induced by the flow of H<sub>2</sub>S through an artificial channel within cement in a high-temperature, high-pressure reactor. Advanced characterization techniques, including CT scanning, SEM, EDS, and FTIR, were employed to analyze the effects of H<sub>2</sub>S alteration on the microstructure, permeability, and chemical composition of the oilwell cement. The main conclusions are as follows.

- (1) The flow of H<sub>2</sub>S through the artificial channel in the oilwell cement leads to an increase in porosity and an expansion of pore sizes within the cement matrix surrounding the channel. This alteration is particularly pronounced near the channel wall. H<sub>2</sub>S induces chemical etching and secondary crystal growth within the cement matrix, causing an enlargement of channel dimensions and increased surface roughness along the channel wall. The effects of H<sub>2</sub>S result in significant swelling and cracking of the cement near the flow inlet, thereby compromising the structural integrity of the cement material.
- (2) The alteration effect of H<sub>2</sub>S significantly increases the permeability of the cement matrix. The average permeability of the cement prior to alteration, under an effective confining pressure of 3.5 MPa, is measured at 0.060 mD. In contrast, the average permeability of cement samples altered by H<sub>2</sub>S for 14 days is recorded at 0.087 mD,

indicating a 45% increase in permeability as a result of H<sub>2</sub>S alteration. This phenomenon can be attributed to H<sub>2</sub>S causing the internal pores of the oilwell cement to expand, thereby increasing porosity and facilitating the initiation of micro-cracks.

- (3) Unaltered oilwell cement displays a uniform texture, characterized by fine particles, low porosity, and a scarcity of large pores at the microscopic level. In contrast, exposure to H<sub>2</sub>S results in the formation of larger pores and cracks within the cement, causing the cement material to exhibit an uneven and fragmented appearance. Additionally, a decalcified, silica-rich layer approximately 1 mm thick forms on the surface of the channel wall as a result of the sulfidation reaction. The detection of metal sulfides in FTIR analyses elucidates the chemical mechanisms underlying the deterioration of the cement's properties.

The findings of this study can be extrapolated to more complex wellbore engineering scenarios, providing important insights for evaluating the corrosion patterns and mechanisms associated with high-temperature H<sub>2</sub>S flow along leaking channels. However, further development of this experiment is necessary. Future analyses should focus on the mechanisms through which H<sub>2</sub>S contributes to the reduction in the strength of oilwell cement. Additionally, subsequent studies should aim to provide insights regarding the influences of temperature and pressure, especially for the conditions of extreme high temperature and high pressure.

#### CRediT authorship contribution statement

**Yue Yin:** Writing – original draft, Methodology, Investigation, Conceptualization. **Li-Wei Zhang:** Writing – review & editing, Supervision, Resources, Methodology, Conceptualization. **Kai-Yuan Mei:** Methodology, Investigation. **Xiao-Wei Cheng:** Supervision. **Man-Guang Gan:** Software. **Yan Wang:** Resources. **Chun-Mei Zhang:** Resources.

#### Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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